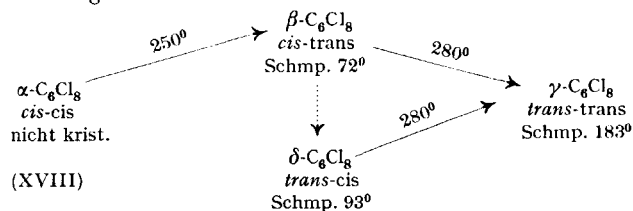


folgende weitere Befunde nahegelegt: α -, β - und δ - C_6Cl_8 werden durch verschiedene Agenzien (Cl_2 , Br_2 , HNO_3) in das am höchsten schmelzende und am schwersten lösliche γ -Isomere übergeführt. Die Umlagerung ist auch durch Erhitzen allein möglich. Bei etwa 250° wandelt sich α - zunächst in β - C_6Cl_8 um, das dann gleichzeitig mit dem δ -Isomeren bei etwa 280° quantitativ in die γ -Form übergeht.

Diesen Beobachtungen wird die zunächst hypothetische Zuordnung der Isomeren nach Schema XVIII am besten gerecht.



Alle bisherigen Versuche machen die Existenz atropisomerer Polychloropolyene höchst wahrscheinlich¹; ein schlüssiger Beweis für eine Atropisomerie kann aber nur mit physikalischen Prüfmethode (UV-Absorption, Raman-Spektren, Dipolmomente usw.) erbracht werden.

Diesbezügliche Untersuchungen an den C_6Cl_8 -Körpern sind in die Wege geleitet worden.

A. ROEDIG

Chemisches Institut der Universität Würzburg, den 27. März 1948.

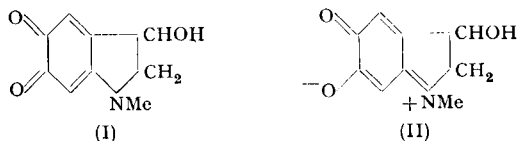
Summary

The synthesis of some unsaturated polychloro compounds, starting from the heptachloropropanes, octachloropropane and hexachloropropane, by means of aluminium in ether is described. For four compounds C_6Cl_8 , subject to some conversions, the structure of octachlorohexatriene-(1,3,5) has been made probable, and the possibility of a special stereoisomerism is discussed.

¹ Bei dem kürzlich von H. GERDING und G. W. A. RIJNDERS (Rec. Trav. chim. Pays-Bas 66, 225 [1947]) raman-spektroskopisch untersuchten Oktachlorpentadien-(1,3) ist trotz behinderter Rotation um die C_2-C_3 -Einfachbindung eine Atropisomerie sterisch unmöglich.

The Structure of Adrenochrome and its Reduction Products

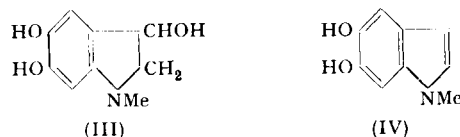
Adrenochrome, the red pigment obtained by the oxidation of adrenalin¹, has generally been formulated as the o-quinone (I). It can also, however be formulated as a zwitterion p-quinoneimine (II), this structure differing from (I) only in electron distribution. It seems probable that this zwitterion mesomeric structure does in fact make a large contribution to the resonance hybrid on the basis of the following observations.



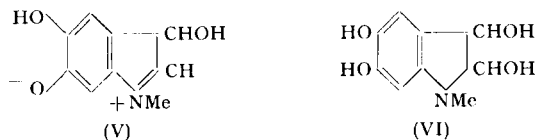
Adrenochrome is very soluble in water, fairly soluble in alcohol but almost insoluble in benzene and ether: behaviour analogous to that of the aliphatic amino-acids and strongly indicative of a zwitterion structure. It has already been observed that adrenochrome readily

forms a monoxime¹ and a mono-semicarbazone. All attempts by the present author to induce reaction with a second molecule of hydroxylamine or semicarbazide have proved unsuccessful, indicating that the molecule probably contains only one true carbonyl group. In general, o-quinones react cleanly with o-phenylenediamine to give phenazine derivatives. Adrenochrome reacts with o-phenylenediamine, but under a wide variety of conditions only ill-defined amorphous products could be obtained and there was no indication of the formation of a phenazine. Adrenochrome is much more intensely coloured than either o-quinone or a quinoneimine salt; however an intensification of absorption would be expected in a molecule which was a resonance hybrid of both these structural types.

The reduction of adrenochrome is of particular interest, and moreover, can only be interpreted satisfactorily on the basis of structure (II). Hitherto it has been supposed that on reduction a leuco-compound (III), re-oxidizable to adrenochrome, is obtained²; though this leuco-compound has in fact never been isolated.



It has now been found that reduction with hydrogen – Pd charcoal in aqueous solution at room temperature and atmospheric pressure presents some rather unusual features and does not give rise to (III). Hydrogen absorption ceases when only one atom of hydrogen per mol. of adrenochrome has been absorbed, and two products are obtained in approximately equal amounts. One of these is 5:6-dihydroxy-N-methylindole (IV), which can easily be extracted with ether, and crystallizes from light petroleum in colourless needles, m.p. $135-136^\circ$. On acetylation it yields the diacetyl derivative m.p. 101° already described³. The other product which is not extracted by ether is the zwitterion (V).



This could not be isolated as such owing to the ease with which its solution decomposed on concentration, but on treatment with alkali 2:3:5:6-tetrahydroxy-N-methyldihydroindole (VI) was readily obtained, the change being analogous to the formation of a pseudobase from a quarternary salt. (VI) forms yellow prisms m.p. 228° dec., very sparingly soluble in water giving yellow solution with a strong green fluorescence, particularly marked in ultra-violet light. Acetylation with acetic anhydride-pyridine gave a dehydration product, 3:5:6-triacetoxy-N-methyl indole (VII), m.p. 111° . It was found by GADDUM and SCHILD⁴ that dilute alkaline solutions of adrenalin absorb oxygen from the air with the development of a green fluorescence and it is suggested that this phenomenon is due to the formation of (VI) by a similar series of reactions. In any case the contention of UTEVSKY⁵ that the fluorescence

¹ D. E. GREEN and D. RICHTER, Biochem. J. 31, 596 (1937).

² Cf. G. B. WEST, Brit. J. Pharm. Chemotherapy 2, 121 (1947).

³ F. BERGEL and A. L. MORRISON, J. Chem. Soc. London, 48 (1943).

⁴ J. H. GADDUM and H. SCHILD, J. Physiol. 89, 9 P (1934).

⁵ UTEVSKY, Advances Med. Biol. 18, 45 (1944).

